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APPLICATION NO.	FILING DATE	FIRST NA	MED INVENTOR		ATTORNEY DOCKET NO.
09/284,160	10/25/99	EYAL		А	U012190-3
Г		! that there is a first	٦.	EXAMINER	
LADAS & PAR	RY	HM12/07	1. 7	OH.T	
26 WEST 61ST STREET				ART UNIT	PAPER NUMBER
NEW YORK NY	' 10023			1623 DATE MAILED:	07/19/01

Please find below and/or attached an Office communication concerning this application or proceeding.

Commissioner of Patents and Trad marks

Office Action Summary

Application No. 09/284,160

Applicant(s)

Eyal et al

Examiner

TAYLOR VICTOR OH

Art Unit 1623



The MAILING DATE of this communication appo	ears on the cover sheet with the correspondence address		
Period for Reply	ALCO TO ALCO T		
A SHORTENED STATUTORY PERIOD FOR REPLY IS THE MAILING DATE OF THIS COMMUNICATION.			
- Extensions of time may be available under the provisions of 37 CF	R 1.136 (a). In no event, however, may a reply be timely filed		
after SIX (6) MONTHS from the mailing date of this communicated. If the period for reply specified above is less than thirty (30) days, a be considered timely.	reply within the statutory minimum of thirty (30) days will		
 If NO period for reply is specified above, the maximum statutory period 	riod will apply and will expire SIX (6) MONTHS from the mailing date of this		
communication Failure to reply within the set or extended period for reply will, by st	atute, cause the application to become ABANDONED (35 U.S.C. § 133).		
 Any reply received by the Office later than three months after the nearned patent term adjustment. See 37 CFR 1.704(b). 	nailing date of this communication, even if timely filed, may reduce any		
Status	,		
1) X Responsive to communication(s) filed on <u>Apr 3</u>	0, 2001		
Zu/ Time delicit is that I	action is non-final.		
3) Since this application is in condition for allowand closed in accordance with the practice under	e except for formal matters, prosecution as to the merits is ix parte Quayle35 C.D. 11; 453 O.G. 213.		
Disposition of Claims			
4) 🗓 Claim(s) <u>19-34</u>	is/are pending in the applica		
4a) Of the above, claim(s)	is/are withdrawn from considera		
5)	is/are allowed.		
	is/are rejected.		
7\\ Claim(s)	is/are objected to.		
(N) Claims	are subject to restriction and/or election requirem		
o) Lidinis			
Application Papers 9) ☐ The specification is objected to by the Examiner.			
10) The drawing(s) filed on	is/are objected to by the Examiner.		
11) The proposed drawing correction filed on	is: a∏ approved h)∏disapproved.		
11) The proposed drawing correction filed on	minor		
12) \square The oath or declaration is objected to by the Exa	milier.		
Priority under 35 U.S.C. § 119			
13) Acknowledgement is made of a claim for foreign	priority under 35 U.S.C. § 119(a)-(d).		
a)☐ All b) ☐ Some* c) ☐None of:			
 Certified copies of the priority documents h 			
Certified copies of the priority documents h	ave been received in Application No		
 Copies of the certified copies of the priority application from the International Bu *See the attached detailed Office action for a list of 	documents have been received in this National Stage reau (PCT Rule 17.2(a)). the certified copies not received.		
*See the attached detailed Office action for a list of 14) ☐ Acknowledgement is made of a claim for domes			
14) Acknowledgement is made of a claim for domes	ino priority under do d.e.e. g		
Attachment(s)			
15) X Notice of References Cited (PTO-892)	18) Interview Summary (PTO-413) Paper No(s).		
16) Notice of Draftsperson's Patent Drawing Review (PTO-948)	19) Notice of Informal Patent Application (PTO-152)		
17) Information Disclosure Statement(s) (PTO-1449) Paper No(s).	20) [] Other:		

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Status of Claims:

Claims 1-18 have been canceled.

Claims 19-34 have been rejected.

1. Applicants' arguments with respect to claims 19-34 have been considered but are moot in

view of the new ground(s) of rejection.

2. The rejection of claims 1 and 18 has been withdrawn due to the modification in the claims.

Claim Rejections - 35 USC § 103

3. This application currently names joint inventors. In considering patentability of the claims

under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was

commonly owned at the time any inventions covered therein were made absent any evidence to

the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor

and invention dates of each claim that was not commonly owned at the time a later invention was

made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35

U.S.C. 102(f) or (g) prior art under 35 U.S.C. 103(a).

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Claim Rejections - 35 USC § 103

- The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 5. The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
 - 1. Determining the scope and contents of the prior art.
 - 2. Ascertaining the differences between the prior art and the claims at issue.
 - 3. Resolving the level of ordinary skill in the pertinent art.
 - 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 6. Claims 19-34 are rejected under 35 U.S.C. 103(a) as being unpatentable over Baniel et al (U.S. 5,510,526) in view of Metz et al (U.S. 4,282,385).

Baniel et al discloses a process for the recovery of lactic acid, from a lactate solution composed of sodium lactate, calcium lactate or potassium lactate (see col. 11, lines 33-34), from a fermentation broth at a pH in the range of 5.5 to 6.5 (see col. 6, lines 6-7) by using a long-chain trialkyl amine in the presence of carbon dioxide by way of extraction (see col. 3, lines 39-44). In the process the organic phase obtained from the primary extraction is further subjected to a separation process such as back extraction, vaporization (see col. 4, lines 60-65) to recover 97 % lactic acid from the original mixture (see col. 11, lines 8-9); the solvent can be used with water

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for the purpose of diluting viscous trialkyl amines or enhancing the extraction (see col. 4, lines 42-46). Also, the reference teaches that it is plausible to recover the lactic acid by acidifying the fermentation broth with sulfuric acid; as a result, a sulfate salt is formed (see col. 1, lines 55-59).

However, the instant invention differs from Baniel et al in that the ratio between free lactic acid and lactate salt is not mentioned.

Metz et al teach a process of manufacturing compounds more than 3 moles of free lactic acid per mole of calcium lactate by using calcium oxide, calcium hydroxide, and calcium carbonate (see col. 2, lines 28-33).

Therefore, if person having an ordinary skill in the art had wished to increase the yield of the lactic acid, it would have been obvious for the skillful artisan in the art to have used Metz et al's ratio between free lactic acid and lactate salt in the fermentation broth in the Baniel et al's process for the recovery of the lactic acid in order to increase the efficiency of the overall process.

The prior art made of record and not relied upon is considered pertinent to applicant's disclosure.

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Eyal et al (U.S. 5,766,439) discloses a process for producing an organic acid in the following steps: producing an organic acid by fermentation, adding an alkaline earth base to the fermentation, reacting the alkaline earth salt of the organic acid with a source of ammonium ions, reducing the concentrations of divalent cations, and converting the ammonium of the organic acid to free organic acid.

Sterzel et al (U.S. 5,453,365) discloses a preparation of lactates by fermentation of the mixture of sugars, conversion of the lactic acid followed by esterification during the process, in which the lactic acid is neutralized with an alkaline earth metal carbonate, added with ammonia and carbon dioxide, and the purified ammonium lactate solution is esterified with an alcohol.

Urbas (U.S. 4,444,881) discloses a process for the recovery of organic acids from dilute aqueous solutions in the following steps: adding a water-soluble tertiary amine carbonate to the calcium salt solution to form the trialkylammonium salt of the acid, and heating the concentrated trialkylammonium salt solution to obtain the acid and the amine.

Cockrem et al (U.S. 5,210,296) discloses a process for producing a high pure lactate ester or lactic acid from a concentrated fermentation broth by acidification in the presence of an alcohol with sequential esterification, distillation of high purity ester.

Walkup et al (U.S. 5,252,473) discloses a process of producing lactic acid and esters of lactic acid in the following reactions. In the first reaction, ammonium lactate produced by a fermentation process of carbohydrate materials can be decomposed into NH₃ and lactic acid; in the second reaction, the lactic acid can be further esterified with methanol to yield methyl lactate.

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In the esterification of the ammonium lactate to the alkyl lactate, the reaction mixture pressure is from 1 atmosphere to 200 atmospheres and the reaction temperature is from 100° C. To 200° C.; besides, the range for the molar ratio of alcohol to ammonium salt in the reaction mixture is from 1:1 to 10:1. In the process, in order to increase the yield of methyl lactate, NH₃ can be either removed or recycled to produce ammonium lactate.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to T. Victor Oh whose telephone number is (703) 305-0809. The examiner can normally be reached on Monday through Friday from 8:30 to 5:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Gary Geist, can be reached on (703) 308-1701. The fax phone number for the organization where this application or proceeding is assigned is (703) 308-4556.

1/11/201

PAUL J. KILLOS

PRIMARY EXAMINER

A.4.1623